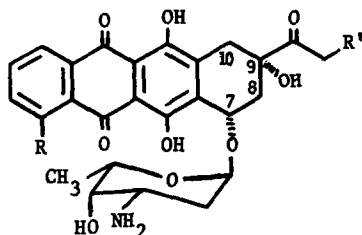


### THE SYNTHESIS OF 4-DEMETHOXYDAUNOMYCIN

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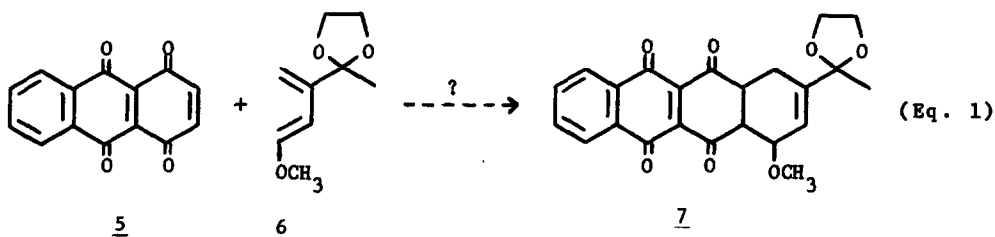
In the past decade the anthracyclines adriamycin (1), daunomycin (2) and carminomycin (3) have come to be recognized as being among the most effective drugs available for the treatment of a broad spectrum of human cancers. More recently, 4-demethoxydaunomycin (4) has been shown



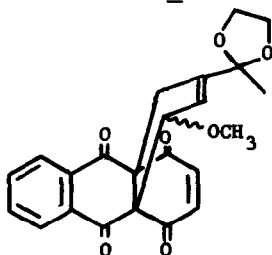
- 1, R = OCH<sub>3</sub>, R' = OH  
2, R = OCH<sub>3</sub>, R' = H  
3, R = OH, R' = H  
4, R = R' = H

to exhibit even greater activity than 1.<sup>2</sup> We now report a synthesis of this compound.<sup>3</sup>

As part of a program directed toward the development of effective routes to the anthracyclines,<sup>4</sup> we had occasion to examine the feasibility of an approach which embodied, as the key constructive step, the Diels Alder reaction between 5<sup>5</sup> and 6 (Eq. 1).<sup>6,7</sup> Distressingly, this

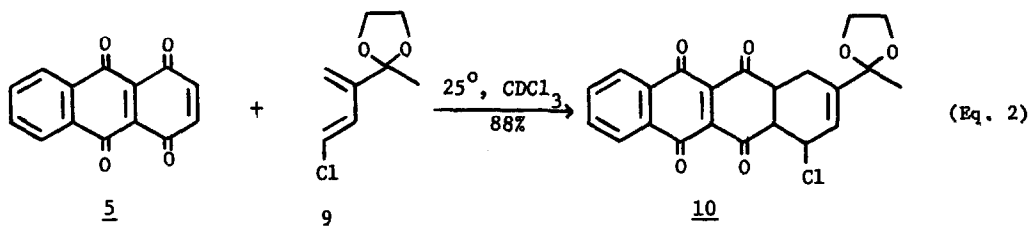


putative route to 7 fails utterly, due to the obstruction of the alternative cycloaddition pathway which affords the "internal" adduct 8 in >80% isolated yield. This finding initially

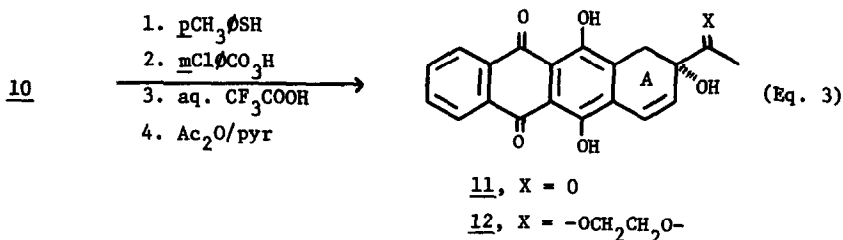


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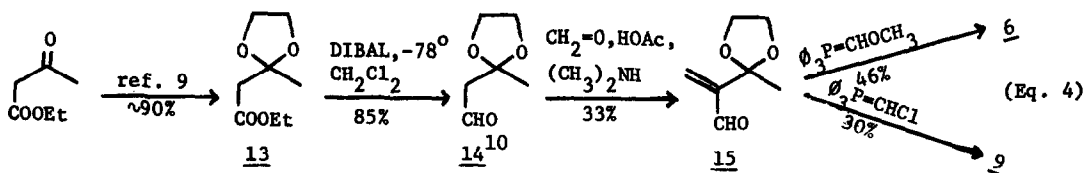
provided a less than welcome incentive to explore alternate approaches to the anthracyclines, but further investigation afforded a less drastic solution. In particular, the simple expedient<sup>8</sup> of replacing methoxydiene 6 with chlorodiene 9 overcomes the disastrous outcome of the original cycloaddition with 5 and gives the desired adduct 10 in 88% yield (Eq. 2).



Adduct 10 can be converted to 11 in four steps (40% overall yield from 5) using the sequence indicated in Equation 3; omission of the ketal hydrolysis step (aq.  $\text{CF}_3\text{COOH}$ ) gives 12 in 51% overall yield from 5.

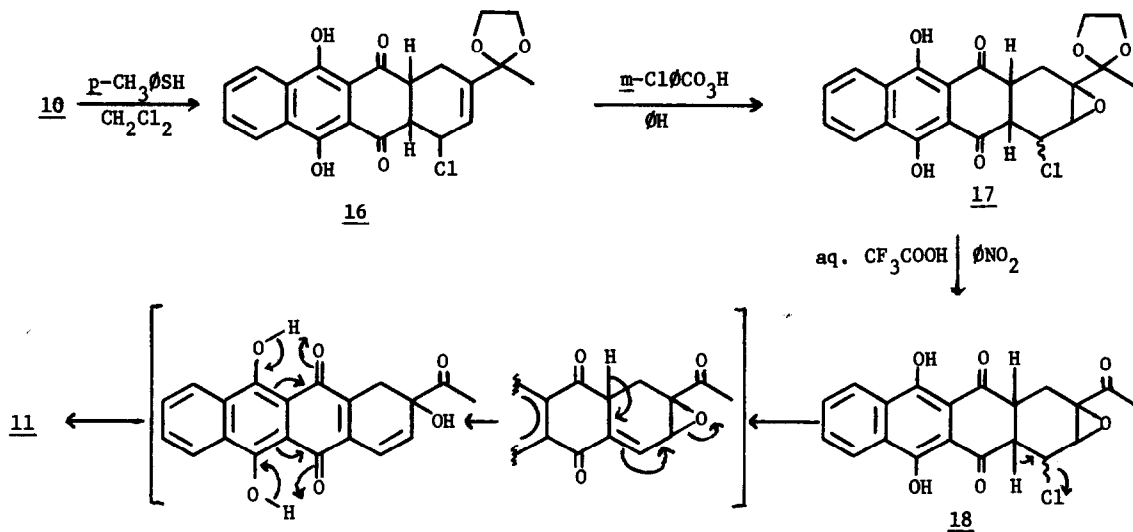


Multigram quantities of dienes 6 and 9 are available in four steps from acetoacetic ester using the incompletely optimized sequences shown in Equation 4.<sup>10</sup>

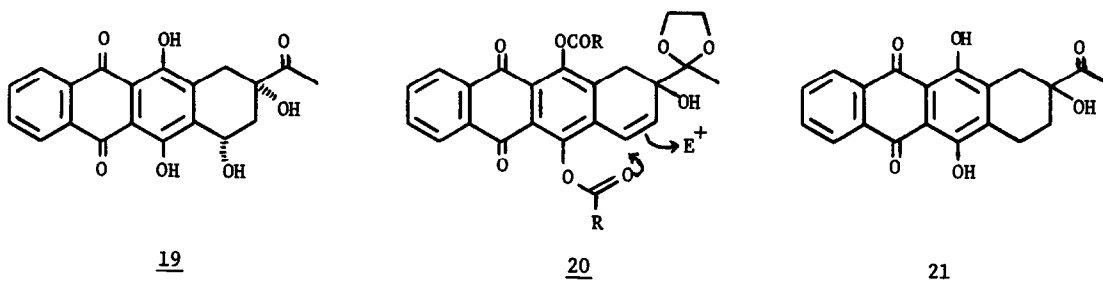


Consideration of the reagents employed in Eq. 3 might suggest that the development of the A-ring functionality proceeds via a 1,3-allylic sulfoxide transposition. In fact, however, characterization of intermediates 16, 17 and 18 reveals the incursion of an entirely different pathway (Scheme 1).

Scheme 1



With 11 and 12 in hand, conversion of these compounds to 4-demethoxydaunomycinone (19) would appear trivial. But innumerable attempts [ $\text{CF}_3\text{COOH}$ ,  $\text{Hg}(\text{OCOCF}_3)_2$ ,  $\text{HX}$ ,  $\text{AcOH/hv}$ , hydrozirconation,  $\text{HOX}$ ,  $\text{OSeO}_2\text{CR}$ , etc., as well as the generalized approach suggested by 20] to achieve this seemingly straightforward transformation so far have been completely unavailing (in most cases A-ring aromatization is the principle mode of reaction observed).



In view of these difficulties, confirmation of the structure of 11 was sought. This was achieved by reduction ( $\text{H}_2$ ,  $\text{Pd/C}$ ,  $\text{EtOH}$ , 1 atm) of 11 to ( $\pm$ )-21. The ( $\pm$ )-21 so obtained (mp  $193\text{--}95^\circ$ )<sup>11</sup> is identical (except for chirality-dependent properties) to an authentic sample of (*S*)-21 provided by Dr. F. Arcamone.<sup>2,11</sup> The obtention of 21 thus affirms the structure assigned to 11; it also constitutes a formal total synthesis of 4-demethoxydaunomycin (4).<sup>2,3b</sup> Efforts to achieve the 11/12  $\rightarrow$  19 transformation continue.<sup>12</sup>

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- For related studies on the Diels Alder reactions of quinizarinquinones (e.g., 5) see H.H. Inhoffen, H. Muxfeldt, V. Koppi and J. Heimann-Trosien, *Ber.*, **90**, 1448 (1957); W.W. Lee, A.P. Martinez, T.H. Smith and D.W. Henry, *J. Org. Chem.*, **41**, 2296 (1976); A.M. Birch, A.J.H. Mercer, A.M. Chippendale, and K.W. Greenhalgh, *J.C.S. Chem. Comm.*, 745 (1977) and ref. 4a. See also J. Altman, E. Cohen, T. Maymon, J.B. Petersen, N. Reshef and D. Ginsburg, *Tetrahedron*, **25**, 5115 (1969).
- For the use of the 5-methoxy derivative of 5 in the synthesis of daunomycinone see: A.S. Kende, Y.-G. Tsay and J.E. Mills, *J. Am. Chem. Soc.*, **98**, 1967 (1976).
- That the substitution of 9 for 6 might lead to the desired terminal adduct (10) was suggested by the observations of Kende *et al.*<sup>7</sup> and ourselves.<sup>4a</sup> Specifically, we envisaged that replacement of 6 with the less electron-rich diene 9 might afford the desired external<sup>4a</sup> adduct 10.
- Cf. E.J. Salmi, *Ber.*, **71B**, 1803 (1938).
- The reduction of 13 to 14 is routinely conducted on a 100g scale. For an alternative synthesis of 14 see T. Oishi, M. Nagai and Y. Ban, *Tetrahedron Lett.*, 491 (1968).
- Racemic 21 has also been prepared by Kende *et al.*<sup>3b</sup> who reported that its mp is 160-62°. The difference in the melting points observed by Kende *et al.* and ourselves is presumably due to polymorphism. The nmr spectrum of "our" 21 is superimposable on that of the (S)-21 provided by Dr. Arcamone (both spectra recorded on our instrument) and identical to that reported for (+)-21 by Kende *et al.*<sup>3b</sup> except that the two proton singlet for the *peri* OH's which Kende *et al.* report<sup>3b</sup> appears at  $\delta$  13.54 occurs at  $\delta$  13.32 on our instrument [for both "our" 21 and the (S)-21].
- Melting points or boiling points (uncorrected) and 60 MHz spectra (in CDCl<sub>3</sub>/TMS, only distinctive peaks are given) for all new compounds are: 6: bp  $\nu$ 55° (.3 mm),  $\delta$  1.50 (3H, s, CH<sub>3</sub>), 3.60 (3H, s, OCH<sub>3</sub>), 3.95 (4H, m, -OCH<sub>2</sub>CH<sub>2</sub>O-), 5.00 (1H, br s, C=CH<sub>b</sub>-H<sub>a</sub>), 5.09 (1H, d, J $\nu$ 2Hz, C=CH<sub>a</sub>-H<sub>b</sub>), 5.42 (1H, d, J=12Hz, CH<sub>3</sub>OCH=CH-), 6.94 (1H, d, J=12Hz, CH<sub>3</sub>OCH=). 8: mp 218-20°,  $\delta$  1.50 (3H, s, CH<sub>3</sub>), 1.95 (1H, br d, J=19Hz, >CH<sub>b</sub>-H<sub>a</sub>), 3.21 (3H, s, OCH<sub>3</sub>), 3.28 (1H, d, J=19Hz, >CH<sub>a</sub>-H<sub>b</sub>), 3.9 (4H, m, -OCH<sub>2</sub>CH<sub>2</sub>O-), 4.77 (1H, d, J=5Hz, CHOCH<sub>3</sub>), 6.30 (1H, m, -CH(OCH<sub>3</sub>)CH=C), 6.56, 6.73, 6.97, 7.14 (2H, ABq, -COCH=CHCO-), 7.95 (4H, m, Ar-H). 9 bp 80° (17 mm, Kugelrohr),  $\delta$  1.52 (3H, s, CH<sub>3</sub>), 3.88 (4H, m, -OCH<sub>2</sub>CH<sub>2</sub>O-), 5.15 (1H, br s, C=CH<sub>b</sub>-H<sub>a</sub>), 5.38 (1H, d, J $\nu$ 2Hz, C=CH<sub>a</sub>-H<sub>b</sub>), 6.28, 6.51, 6.58, 6.81 (2H, ABq, C1CH=CH-). 10: mp 254-50° (starts to turn red at  $\nu$ 1300°),  $\delta$  1.55 (3H, s, CH<sub>3</sub>), 3.98 (4H, br s, -OCH<sub>2</sub>CH<sub>2</sub>O-), 4.94 (1H, br t, J $\nu$ 5Hz, C-7 H), 6.10 (1H, br d, J $\nu$ 5Hz, C-8 H), 7.9 (4H, m, Ar-H). 11: mp 230-33° (insufficiently soluble for useful NMR spectrum), 12: mp 208-10°,  $\delta$  1.40 (3H, s, CH<sub>3</sub>), 2.92, 3.22, 3.29, 3.59 (2H, ABq, C-10 H's), 4.07 (4H, s, -OCH<sub>2</sub>CH<sub>2</sub>O-), 6.26, 6.41, 7.07, 7.22 (2H, ABq, C-7, C-8 H's), 7.78 (2H, m, Ar-H), 8.30 (2H, m, Ar-H), 13.27 (1H, s, ArOH), 13.37 (1H, s, ArOH). 14: bp 70-73° (16 mm) [lit.<sup>10</sup> 68-69.5 (11 mm)],  $\delta$  1.40 (3H, s, CH<sub>3</sub>), 2.67 (2H, d, J=3Hz, CH<sub>2</sub>CHO), 3.99 (4H, s, -OCH<sub>2</sub>CH<sub>2</sub>O-), 9.75 (1H, t, J=3Hz, -CHO). 15: bp 93° (15 mm),  $\delta$  1.57 (3H, s, CH<sub>3</sub>), 3.91 (4H, m, -OCH<sub>2</sub>CH<sub>2</sub>O-), 6.12 (1H, d, J=1Hz, C=CH<sub>b</sub>-H<sub>a</sub>), 6.51 (1H, d, J=1Hz, C=CH<sub>a</sub>-H<sub>b</sub>), 9.65 (1H, s, -CHO). 16: mp 184-88°,  $\delta$  1.60 (3H, s, CH<sub>3</sub>), 3.97 (4H, m, -OCH<sub>2</sub>CH<sub>2</sub>O-), 4.95 (1H, m, C-7 H), 6.19 (1H, m, C-8 H), 7.77 (2H, m, Ar-H), 8.58 (2H, m, Ar-H), 12.98 (1H, s, ArOH), 13.67 (1H, s, ArOH). 17: mp 218-20°,  $\delta$  1.55 (3H, s, CH<sub>3</sub>), 3.99 (4H, s, -OCH<sub>2</sub>CH<sub>2</sub>O-), 4.79 (1H, vt, J $\nu$ 3Hz, C-7 H), 7.75 (2H, m, Ar-H), 8.45 (2H, m, Ar-H), 13.07 (1H, s, ArOH), 13.75 (1H, s, ArOH). 18: mp 225-80° (starts to turn red  $\nu$ 1300°),  $\delta$  (CF<sub>3</sub>COOH/TMS) 2.44 (3H, s, COCH<sub>3</sub>), 3.97 (3H, m), 5.06 (1H, m, C-7 H), 7.96 (2H, m, Ar-H), 8.61 (2H, m, Ar-H).